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## **Chapter 13 Section 1 Review Solutions**

CHAPTER 13 REVIEW

Solutions SECTION

13-2 SHORT ANSWER

Answer the following questions in the space provided. 1. Following are statements about the dissolving process. Explain each one at the molecular level. a. Increasing the pressure of a solute gas above a

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## Solutions Section

liquid solution

increases the solubility  
of the gas

### **CHAPTER 13 REVIEW Solutions**

This Chapter 13  
Review, Section 1: Ions  
in Aqueous Solutions  
and Colligative  
Properties Worksheet is  
suitable for 7th - 12th  
Grade. This is an apt  
assignment for  
chemistry takers that  
are studying ionic  
solutions. Eight

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## Solutions Section

questions require  
problem solving and  
critical thinking to  
answer.

### **Chapter 13 Review, Section 1: Ions in Aqueous Solutions**

...

Chapter 13 Review  
Answer Key

Understanding

Vocabulary Section

13.1 1. electric current

2. electrical symbols 3.

switch 4. open circuit

5. closed circuit 6.

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Solutions Section  
Answers

resistance Section 13.2  
7. ampere 8. battery 9.  
voltage 10. volt Section  
13.3 11. ohm 12.  
Ohm's law 13.  
resistance 14.  
potentiometer 15.  
conductor Reviewing  
Concepts Section 13.1  
1 ...

**Chapter 13 Review  
Answer Key -  
Northern Highlands  
Chemistry and  
Chemical Reactivity  
(9th Edition) answers**  
*Page 8/26*



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to Chapter 13 Solutions  
and Their Behavior -

13-1 Units of  
Concentration - Review  
& Check for Section  
13-1 - Page 473 1

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community members  
like you. Textbook

Authors: Kotz, John C.;

Treichel, Paul M.;

Townsend, John R.;

Treichel, David A.,

ISBN-10: 1133949649,

ISBN-13:

978-1-13394-964-0,

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### **Chemistry and Chemical Reactivity (9th Edition)**

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Terms in this set (27)

Material doing the  
dissolving. solvent.

Material being

dissolved. solute. A

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Solutions Section  
1 Answer

homogeneous mixture  
of two or more

substances. solution.  
solution of metals.

alloy. what is the rule  
for selecting a solvent.  
like dissolves like.

Which solvent  
discussed in ...

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13 Solutions Chapter  
13 - Section 1

Objectives: 1.

Distinguish between  
heterogeneous and  
homogeneous

mixtures. 2. List three

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Solutions Section  
1 Answer  
different solutesolvent  
combinations. 3.

Compare the  
properties of  
suspensions, colloids,  
and solutions. 4.

Distinguish between  
electrolytes and  
nonelectrolytes.

## **CHEMISTRY -**

### **Chapter 13**

CHAPTER 12 REVIEW

Solutions SECTION 2

SHORT ANSWER

Answer the following  
questions in the space

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## Solutions Section Answers

provided. 1. The following are

statements about the dissolving process.

Explain each one at the molecular level. a.

Increasing the pressure of a solute gas above a liquid solution increases the solubility of the gas

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Modern Chemistry 7  
Solutions CHAPTER 12  
REVIEW Solutions

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## Solutions Section SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. a. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid.

## **CHAPTER 12 REVIEW** **Solutions**

This Chapter 13  
*Page 15/26*

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Solutions Section  
Review, Mixed Review:

Ions in Aqueous

Solutions and

Colligative Properties

Worksheet is suitable

for 9th - 12th Grade.

Here is an attractive

worksheet that walks

chemistry learners

through a review of

aqueous solutions.

There are matching,

short answer, and

multiple choice

questions dealing with

boiling and freezing

points, precipitate,



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Solutions Section  
1. Molality, net ionic  
equations, and ...

## **Chapter 13 Review, Mixed Review: Ions in Aqueous Solutions**

...

CHAPTER 13 REVIEW  
Solutions MIXED  
REVIEW SHORT

ANSWER Answer the following questions in the space provided. 1. Solid  $\text{CaCl}_2$  does not conduct electricity, but it is considered to be an electrolyte. Explain.

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Solutions Section  
4.13.1

2. Explain the following statements at the molecular level: a. Generally a polar liquid and a nonpolar liquid are immiscible.

## **HOMEWORK 13-1**

This video explains the concepts from your packet on Chapter 13 (Properties of Solutions), which can be found here:

<https://goo.gl/etUYcM>

Section 13.1: The Solution Process

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13.2: Saturated  
1..Answers

**Chapter 13**  
**Properties of**  
**Solutions**

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This Chapter 13 Review, Section 2: Ions in Aqueous Solutions and Colligative Properties Worksheet is suitable for 9th - 12th Grade. Here is a general chemistry instructional activity that incorporates practical applications. Four questions employ critical thinking about solutions, boiling and freezing points, and molar mass.

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## **Chapter 13 Review, Section 2: Ions in Aqueous Solutions**

...

Chapter 13: Standard Review Worksheet 1.  
While the barometer is used to measure atmospheric pressure, a device called a mercury manometer is used to measure the pressure of samples of gas in the laboratory. A manometer consists basically of a U-shaped

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Solutions Section

tube filled with  
mercury, with one arm  
of the

## **Chapter 13: Standard Review Worksheet**

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Section Review 13.2  
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your Parent/guardian  
grades

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Chapter 13 Review

Exercise Solutions

R13.1 Recursion

Recursion solves a

problem by using the

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## Solutions Section

solution of the same problem with simpler values. Iteration Iteration uses some form of a loop (for loop, while loop) to solve a problem. Infinite recursion Infinite recursion occurs when the recursion never stops.

### **Chapter 13 Review Solutions - Chapter 13 Review Exercise**

...

Start studying

Page 24/26



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13.2. Learn vocabulary, terms, and more with flashcards, games, and other study tools.

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